## Novel $CuX_2$ -mediated cyclization of acid–base salts of (L)-cinchonidine or (D)-/(L)- $\alpha$ -methylbenzylamine and 2,3-allenoic acids in an aqueous medium. An efficient entry to optically active $\beta$ -halobutenolides

## Shengming Ma\* and Shulin Wu

Laboratory of Organometallic Chemistry, Shanghai Institute of Organic Chemistry, Chinese Academy of Sciences, 354 Fenglin Lu, Shanghai 200032 People's Republic of China. E-mail: masm@pub.sioc.ac.cn

Received (in Cambridge, UK) 1st November 2000, Accepted 16th January 2001 First published as an Advance Article on the web 15th February 2001

The treatment of 1:1 salts of 2,3-allenoic acid–chiral base with  $CuX_2$  (4 equiv.) in an aqueous medium, *i.e.* acetone– $H_2O$  (2:1), at 60–65 °C afforded  $\beta$ -halobutenolides with high enantiopurities in good to excellent yields.

Polysubstituted butenolides are a class of compounds of current interest due to their potential broad range of biological activities  $^1$  and abundant occurrence in natural products.  $^2$  However, the methods for the highly stereoselective synthesis of optically active butenolides are limited.  $^{3,4}$  In this paper, we wish to report a highly efficient  $CuX_2$ -mediated cyclization of the salts formed between chiral bases and 2,3-allenoic acids. The method provides a novel route to  $\beta$ -halobutenolides with high enantiopurity, important building blocks for polysubstituted butenolides.  $^5$ 

Recently, we have developed several methodologies for the synthesis of  $\beta$ -halobutenolides from 2,3-allenoic acids.<sup>5,6</sup> The interesting point of these reactions is that the starting 2,3-allenoic acids are a class of compounds with chirality when properly substituted. Thus, it would be possible to use a cheap optically active base to resolve 2,3-allenoic acids and transfer the axial chirality in allenes into central chirality in butenolides in a highly stereoselective manner. One major issue here is the use of the salt of an optically active base with 2,3-allenoic acids directly as the starting point, the release of 2,3-allenoic acids from the salts would not be necessary, which makes this strategy more attractive.

The resolution of racemate 2-methyl-4-phenylbuta-2,3-dienoic acid  ${\bf 1a}$  with 0.5 equiv. of (L)-cinchonidine, a readily available and relatively cheap base, afforded a salt which could be readily recrystallized in *ethyl acetate* to afford the optically active salt (+)- ${\bf 2a}$  in 43% yield with  $[\alpha]_{\rm D}^{20}=+85.4^{\circ}.^7$  Release of the acid from the corresponding salt (+)- ${\bf 2a}$  by the treatment with dilute  ${\bf H}_2{\bf SO}_4$  afforded  ${\it S}$ -(+)- ${\bf 1a}$ , indicating the ( ${\it S}$ )-configuration of the allene moiety according to the Lowe–Brewster rule (Scheme 1).8

Luckily, when (+)-2a was treated with CuBr<sub>2</sub> in an aqueous medium (acetone– $H_2O$  (2:1)), at 60–65 °C for 3 h, a methodology recently developed by ourselves for the halolacto-

Ph Me racemate 
$$CO_2H$$
 Ph  $CO_2^-$  (L)-cinchonidine-H  $^+$  (L)-cinchonidine  $H^+$   $H^ H^ H^-$ 

Scheme 1

nization of 2,3-allenoic acids,<sup>6</sup> the reaction afforded (+)-**3a** in 95% yield with 98% ee, <sup>9</sup> the corresponding  $\beta$ -chlorobutenolide (+)-**3b** was also obtained in 90% yield with 99% ee by using CuCl<sub>2</sub> instead of CuBr<sub>2</sub> (Scheme 2).

Ph Me 
$$CO_2^-$$
 (L)-cinchonidine-H  $^+$   $CUX_2$   $ACCO_2^ ACCO_2^-$  (L)-cinchonidine-H  $^+$   $ACCO_2^ ACCO_2^ ACCO$ 

With the standard aqueous reaction conditions in hand, a series of β-bromobutenolides with high optical purity were prepared and the results are summarized in Table 1. It is obvious: (1) the yields are from good to excellent; and (2) the efficiency of the chirality transfer process is almost 100% since the %ee of the products from the resolved salts are similar to those from the released free 2,3-allenoic acids (Scheme 3), indicating that the chirality of (L)-cinchonidine has almost no impact on the chirality transfer of the allene moiety. Similar results were obtained for all substrates using CuCl2 in place of CuBr<sub>2</sub> to afford β-chlorobutenolides. The absolute configuration of the chiral centers in the products 3c and 3d were determined by X-ray diffraction using the bromine atoms as the reference. 10 The absolute configuration of other products are based on these X-ray studies and further confirmed by the study of their CD spectra.11

Furthermore, it is interesting to observe when we used (S)-(+)- $\alpha$ -methylbenzylamine as the resoluting agent, the salt (—)- $\mathbf{4a}$  was obtained and its treatment with CuBr<sub>2</sub> afforded the opposite enantiomers (R)-(—)- $\mathbf{3a}$  (98% ee) and (R)-(—)- $\mathbf{3b}$  (98% ee) in 90% and 93% yields, respectively (Scheme 4).

By using (R)-(-)- $\alpha$ -methylbenzylamine instead of its S-enantiomer, the corresponding salt (+)- $\mathbf{4a}$  afforded the same enantiomer as with (L)-cinchonidine, *i.e.* (S)-(+)- $\mathbf{3a}$  and (S)-(+)- $\mathbf{3b}$  in 92 (98% ee) and 90% yield (97% ee), respectively (Scheme 4).

In conclusion, we have developed an efficient aqueous synthesis of highly optically active  $\beta$ -halobutenolides. The current methodology will show its utility in organic synthesis due to the ready availability of starting materials with different substitution patterns, 7,12 direct cyclization from the salts, and availability of both enantiomers.

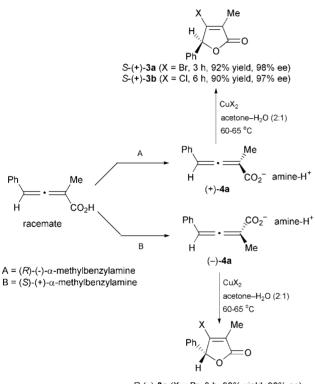
DOI: 10.1039/b008818h Chem. Commun., 2001, 441–442 **441** 

$$R^{1} = R^{3} \qquad \text{or} \qquad R^{2} \qquad R^{3} \qquad \text{or} \qquad R^{2} \qquad R^{3} \qquad \text{base} = (L)-cinchonidine}$$

$$R^{1} = R^{3} \qquad R^{3}$$

Entry	Acid			Optical			
	R <sup>1</sup>	R <sup>2</sup>	R <sub>3</sub>	rotation of the salt <sup>b</sup>	$CuX_2$ X =	Reaction time/h	Yield (%) (product, ee% <sup>9</sup> )
1	C <sub>6</sub> H <sub>13</sub>	Н	Me (1b)	-9.10	Cl	6	90 (S-(-)- <b>3c</b> , 94)
2	$C_6H_{13}$	Н	Me (1b)	-9.10	Br	3	95 $(S-(-)-3d,^c 95)$
3	$C_{10}H_7^d$	Н	Me (1c)	+58.3	Cl	10	88 (S-(+)- <b>3e</b> , 98)
4	$\mathrm{C}_{10}\mathrm{H}_7{}^d$	Н	Me (1c)	+58.3	Br	3	86 (S-(+)- <b>3f</b> , e 98)
5	Ph	Н	$C_3H_7$ (1d)	+37.8	Cl	8	92 (S-(+)- <b>3g</b> , 96)
6	Ph	Н	$C_3H_7$ (1d)	+37.8	Br	3	96 (S-(+)- <b>3h</b> , 96)

reaction was carried out using the salt and  $CuX_2$  (4 equiv.) in acetone– $H_2O$  (2:1) at 60–65 °C. b The specific optical rotation  $[\alpha]_D^{20}$  (c=1, c The absolute configuration was determined by X-ray diffraction studies. d  $C_{10}H_7=\alpha$ -naphthyl.



R-(-)-3a (X = Br, 3 h, 90% yield, 98% ee) R-(-)-3b (X = Cl, 6 h, 93% yield, 98% ee) Scheme 4

Financial support from National NSF of China (No. 29932020) and the Major State Basic Research Development Program (Grant No. G2000077500) are greatly appreciated. Shengming Ma is the recipient of 1999 Qiu Shi Award for Young Scientific Workers issued by Hong Kong Qiu Shi Foundation of Science and Technology (1999–2002) and the Special Starting Grant for Outstanding Young Chemists issued by National Natural Science Foundation of China (29525202).

## Notes and references

(a) T. S. Brima, US 4,968,817, 1990 [Chem. Abstr., 1991, 114, 185246y];
 (b) A. Tanabe, Jpn. Kokai. Tokyo. Koho JP. 63,211,276 [88,211,276], 1988 [Chem. Abstr. 1989, 110, 94978q];
 (c) G. C. M. Lee, Eur. Pat. EP. 372,940, 1990 [Chem. Abstr., 1990, 113, 191137j];
 (d) Y.

- Ducharme, J. Y. Gauthier, P. Prasit, Y. Leblanc, Z. Wang, S. Leger and M. Thrien, PCT Int. Appl. WO 95, 00,501, 1995 [*Chem. Abstr.*, 1996, **124**, 55954y]; (*e*) C. M. Lee Gary and M. E. Gast, PCT Int. Appl. WO. 91 16,055, 1991 [*Chem. Abstr.*, 1992, **116**, 59197m].
- 2 For recent examples, see: (a) Y. Chia, F. Chang and Y. Wu, Tetrahedron Lett., 1999, 40, 7513; (b) S. Takahashi, K. Maeda, S. Hirota and T. Nakata, Org. Lett., 1999, 1, 2025; (c) D. A. G. Cortez, J. B. Fermandes, P. C. Vieria, M. F. Das, G. F. Da Silva, A. G. Ferreira, Q. B. Cass and J. R. Pirani, Phytochemistry, 1998, 49, 2493; (d) H. Ostuka, K. Kotani, M. Bando, M. Kido and Y. Takeda, Chem. Pharm. Bull., 1998, 46, 1180; (e) T. Ishikawa, K. Nishigaya, H. Uchikoshi and I. Chen, J. Nat. Prod., 1998, 64, 534; (f) S. Driol, F. Felluga, C. Forzeto, P. Nitti, G. Pitacco and E. Valentin, J. Org. Chem., 1998, 63, 2385.
- 3 For a summary of methodologies for the synthesis of racemic butenolides, see: (a) D. W. Knight, Contemp. Org. Synth., 1994, 287; (b) S. Ma and Z. Shi, J. Org. Chem., 1998, 63, 6387, and the references cited therein. For some of the most recent examples, see: (c) N. Chatani, T. Morimoto, Y. Fukumoto and S. Murai, J. Am. Chem. Soc., 1998, 120, 5335; (d) M. J. Bassindalem, P. Hamley, A. Leitner and J. P. A. Harrity, Tetrahedron Lett., 1999, 40, 3247.
- 4 For some of the representative examples of the stereoselective synthesis of optically active butenolides, see: (a) A. G. Schultz, M. Dai, S. Khim, L. Pettus and K. Thakkar, *Tetrahedron Lett.*, 1998, **39**, 4203; (b) S. M. Dankwardt, J. W. Dankwardt and R. H. Schlessinger, *Tetrahedron Lett.*, 1998, **39**, 4971, 4975 and 4979; (c) A. van Oeveren and B. L. Feringa, *J. Org. Chem.*, 1996, **61**, 2920; (d) M. Renard and L. Ghosez, *Tetrahedron Lett.*, 1999, **40**, 6237; (e) T. Mukaiyama and K. Suzuki, *Chem. Lett.*, 1980, 255; (f) S. Tsuboi, J. Sakamoto, H. Yamashita, T. Sakai and M. Utaka, *J. Org. Chem.*, 1998, **63**, 1102; (g) Q. Yu, Y. Wu, L.-J. Xia, M.-H. Tang and Y.-L. Wu, *Chem. Commum.*, 1999, 129; (h) T. Berkenbusch and R. Bruckner, *Tetrahedron*, 1998, **54**, 11471.
- 5 (a) S. Ma, Z. Shi and Z. Yu, Tetrahedron Lett., 1999, 40, 2393; (b) S. Ma, Z. Shi and Z. Yu, Tetrahedron, 1999, 55, 12137.
- 6 S. Ma and S. Wu, J. Org. Chem., 1999, 64, 9314.
- 7 The known resolving process in acetone lacks efficiency and affords the salts in 27.5% yield with  $[\alpha]_D^{25} = +60^\circ$ , see: W. Runge and G. Kresze, *Liebigs Ann. Chem.*, 1975, 1361.
- 8 S. R. Landor, *The Chemistry of the Allenes*, Academic Press, New York, 1982, vol. 3, pp. 587–590.
- 9 The ee values were determined by HPLC on a Chiralpack AS column using *n*-hexane–isopropyl alcohol (65:35) as the eluent. The racemic butenolides were prepared by starting from the racemic 2,3-allenoic acids using the method reported in ref. 6.
- 10 The absolute configuration was determined using Texsan software. The Bijvoet reflections were collected and refined with Bijvoets not flagged as redundant. CCDC 152077 and 152078. See http://www.rsc.org/ suppdata/cc/b0/b008818h/ for crystallographic data in .cif format.
- 11 (a) J. K. Gawronski, A. van Oeveran, H. van der Deen, C. W. Leung and B. L. Feringa, J. Org. Chem., 1996, 61, 1513; (b) J. K. Gawronski, Q. Chen, Z. Geng, B. Huang, M. R. Martin, A. I. Mateo, M. Brzostowska, U. Rychlewska and B. Feringa, Chirality, 1997, 9, 537.
- (a) H.-J. Bestmann and H. Hartung, Chem. Ber., 1966, 99, 1198; (b)
   H. D. Venkruijsse and L. Brandsma, Synthesis of Acetylenes, Allenes and Cumulenes. A Laboratory Manual, Elsevier: Amsterdam, The Netherlands, 1981, p. 33.